

STEREOCONTROLLED SYNTHESIS OF (\pm)-RECIFEIOLIDE VIA ORGANOPALLADIUM CHEMISTRY

Barry M. Trost and Thomas R. Verhoeven
Department of Chemistry
University of Wisconsin-Madison
Madison, Wisconsin 53706

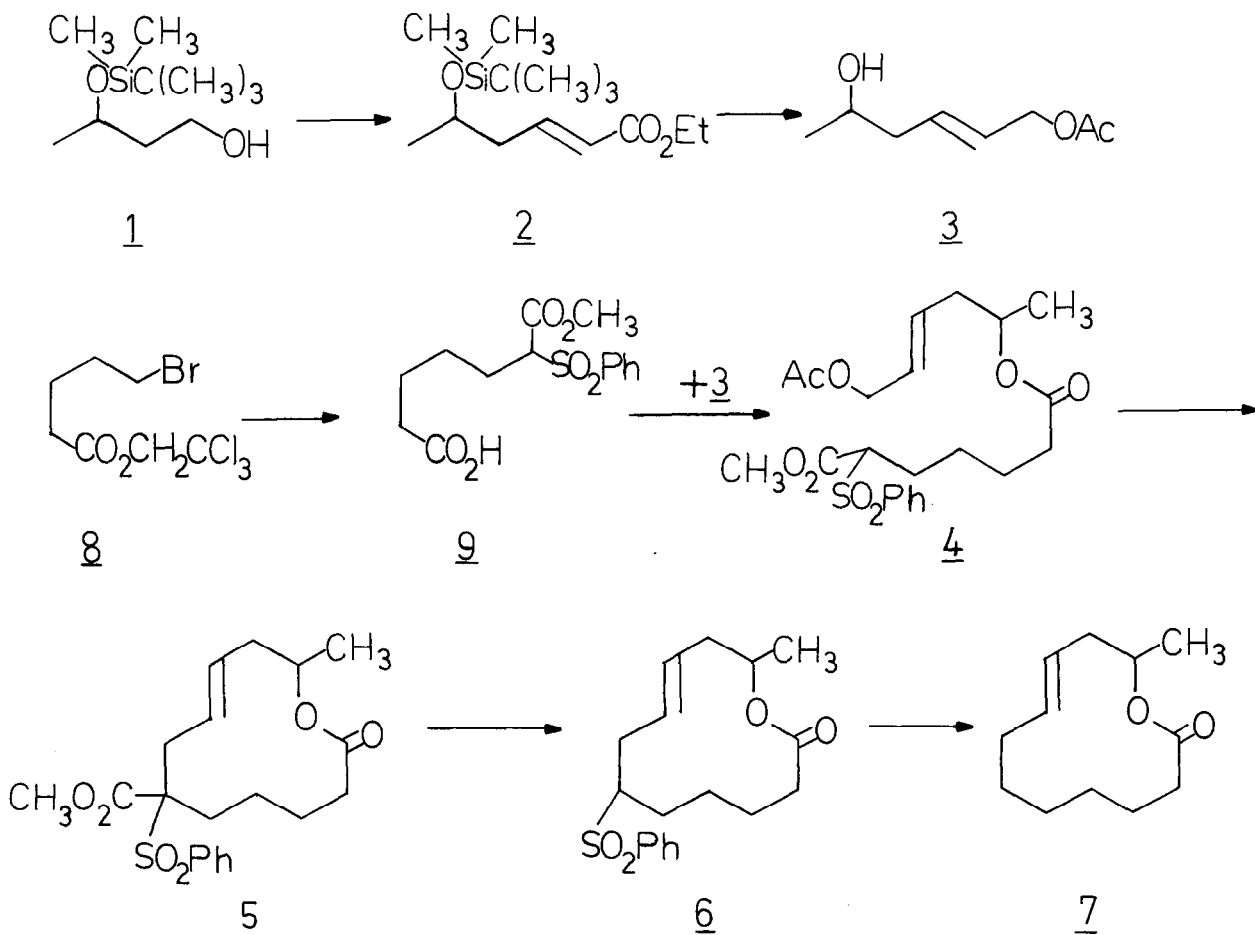
(Received in USA 23 February 1978; received in UK for publication 2 May 1978)

While impressive strides have been made in macrolide synthesis via macrolactonization routes¹, a general method for preparation of macrolides via C-C bond fusion has to date been lacking^{1,2}. The preparation of lactones via C-C bond formation rather than the more conventional lactonization routes potentially provides a greater degree of flexibility in devising synthetic strategies directed toward the growing number of naturally occurring macrolides³. In particular, one may employ a convergent synthetic scheme, allowing synthesis of alcohol and acid portions independently which can then be readily condensed, forming the ester linkage prior to ring closure. A general macrolactonization method recently developed and utilized by ourselves⁴ and others⁵ involves the use of organopalladium chemistry. We now wish to report the application of this methodology in the total synthesis of (\pm)-recifeiolide⁶, a naturally occurring macrolide isolated⁷ from the fungus *Cephalosporium recifei*.

The alcohol portion 3 was readily synthesized starting from tert-butyldimethylsilyl ether 1. Oxidation with pyridinium chlorochromate (2 equiv., NaOAc, CH₂Cl₂, R.T.), followed by an Emmons-Wadsworth-Horner condensation ((EtO)₂POCH₂CO₂Et, NaH, THF, -78°-0°C) afforded the E- α,β unsaturated ester 2. Reduction with diisobutylaluminum hydride (toluene, 0°C) followed by acylation (AcCl, pyridine, CH₂Cl₂, 0°C) and hydrolysis of the silyl ether (HClO₄, THF-H₂O, R.T.) afforded 3⁸ in an overall 53% yield from 1. Acid portion 9 was available in two steps from the β,β,β -trichloroethyl ester of 5-bromo-pentanoic acid, 8. Alkylation of the sodium salt of methyl phenylsulfonylacetate (DMSO, 50°C) and reductive hydrolysis (Zn, DMF, 0°C)⁹ gave acid 9 in 84% yield. Conversion of 9 to its acid chloride (SOCl₂) and condensation with alcohol 3 (Et₂O, pyridine 0°-40°C) gave the requisite precursor 4⁸ in 79% yield. Conversion of 4 to the anion with NaH in THF and addition of the resultant solution via a syringe pump to a refluxing solution of 9 mole % Pd(Ph₃P)₄ produced only the 12-membered lactone 5⁸ (m.p. 126-129.5°C) in 78% yield. Examination of the NMR spectrum at 270MHz revealed the presence of a trans double bond (J = 15.5 Hz), and absence of any terminal vinyl group which would have resulted from reaction at the allylically related carbon. In addition to the first example of 12 membered ring formation, the complete regio-

(no ten membered ring) and stereospecificity (no Z isomer) of this cyclization is quite noteworthy--especially compared to the intermolecular version¹⁰. Completion of the synthesis was accomplished by decarbomethoxylation to 6 ($(\text{CH}_3)_4\text{NOAc}$, HMPA, 95°C) (86%) followed by reductive desulfonylation (6% Na-Hg, Na_2HPO_4 , EtOH-THF, -25°C)¹¹ (94%) to give (\pm)-recifeiolide, 7⁸, identical by spectroscopic (ir, nmr, ms) and chromatographic (tlc, vpc) criteria to an authentic sample¹².

The high regio- and stereospecificity of this cyclization approach thus provides a viable alternative for the formation of medium and large ring lactones. In addition, the sulfone-ester moiety present in the initially cyclized product may also serve to introduce further functionalization in the molecule. The consequences of this are actively being pursued in our laboratories.



Acknowledgement: We wish to thank the donors of the Petroleum Research Fund, administered by the American Chemical Society, the National Science Foundation, and the National Institutes of Health, General Medical Sciences Institute for their generous support of our programs.

REFERENCES

1. For recent reviews see: T. G. Back, Tetrahedron, **33**, 3041 (1977); K. C. Nicolaou, Tetrahedron, **33**, 683 (1977).
2. E. J. Corey and M. B. Petrizilka, Tetrahedron Letters, 2537 (1975); D. Bichan and M. Winnik, Tetrahedron Letters, 3857 (1974); R. N. Hurd and D. H. Shah, J. Org. Chem., **38**, 390 (1973); E. J. Corey and H. A. Kirst, J. Amer. Chem. Soc., **94**, 667 (1972); P. M. Baker, B. W. Bycroft and J. C. Roberts, J. Chem. Soc. (C), 1913 (1967); L. D. Bergelson, J. G. Molotkovsky, and M. M. Shemyakin, Chem. Ind., 558 (1960).
3. W. Keller-Schierlein, Fortschr. Chem. Org. Naturst., **30**, 313 (1973); W. D. Celmer, Pure Appl. Chem., **28**, 413 (1971).
4. B. M. Trost and T. R. Verhoeven, J. Amer. Chem. Soc., **99**, 3867 (1977).
5. Y. Kitagawa, A. Itoh, S. Hashimoto, H. Yamamoto and H. Nozaki, J. Amer. Chem. Soc., **99**, 3864 (1977).
6. For previous synthesis see: E. J. Corey, P. Ulrich, and J. M. Fitzpatrick, J. Amer. Chem. Soc., **98**, 222 (1976); H. Gerlach, K. Oertle and A. Thalman, Helv. Chim. Acta, **59**, 755 (1976); K. Narasaka, M. Yamaguchi and T. Mukaiyama, Chem. Letters, 959 (1977); K. Utimoto, K. Uchida, M. Yamaya and H. Nozaki, Tetrahedron Letters, 3641 (1977).
7. R. F. Vesonder, F. H. Stodola, L. J. Wickerham, J. J. Ellis and W. K. Rohwedder, Can. J. Chem., **49**, 2029 (1971).
8. $^1\text{H-NMR}$ (CDCl_3): δ 5.79 (d of t, J = 15.4, 6.9 Hz, 1H), 5.65 (d of t, J = 15.4, 6.1 Hz, 1H), 4.53 (d, J = 6.1 Hz, 2H), 3.85 (m, 1H), 2.72 (b, 1H), 2.23 (bt, J = 6.6 Hz, 2H), 2.07 (s, 3H), 1.19 (d, J = 6.1 Hz, 3H).
- 4: $^1\text{H-NMR}$ (CDCl_3): δ 7.87 (bd, J = 7 Hz, 2H), 7.70 (bt, J = 7 Hz, 1H), 7.57 (t, J = 7 Hz,

2H), 5.65 (m, 2H), 4.92 (m, 1H), 4.50 (d, J = 5 Hz, 2H), 3.96 (d of d, J = 10.5, 5 Hz, 1H), 3.65 (s, 3H), 2.28 (m, 4H), 2.05 (s, 3H), 1.98 (m, 2H), 1.61 (m, 2H), 1.35 (m, 2H), 1.18 (d, J = 6 Hz, 3H).

5: $^1\text{H-NMR}$ (CDCl_3): δ 7.79 (bt, J = 6.8 Hz, 2H), 7.67 (m, 1H), 7.56 (m, 2H), 5.88 (d of d of t, J = 15.5, 11, 2 Hz, 0.4H), 5.56 (d of d of d of d, J = 15.3, 11, 3.5, 2 Hz, 0.6H), 5.37 (d of d of d of d, J = 15.5, 10.5, 4, 2 Hz, 0.4H), 5.20 (m, 1H), 5.13 (d of d of t, J = 15.3, 11, 2 Hz, 0.6 H), 3.62 (s, 3H), 3.03 (m, 1H), 2.80 (d of d, J = 13.6, 11.0 Hz, 0.6H), 2.55 (d of d, J = 16.5, 11.5 Hz, 0.4H), 2.47-2.14 (m, 5H), 1.43 (m, 2H), 1.27 (d, J = 6.4 Hz, 0.4H), 1.23 (d, J = 6.4 Hz, 0.6H).

7: $^{13}\text{C-NMR}$ (CDCl_3): δ 173.2, 133.4, 127.1, 68.5, 41.1, 33.0, 30.4, 25.1, 24.8, 24.3, 23.4, 20.6.

All nmr spectra were recorded at 270 MHz. All new compounds have been fully characterized including determination of elemental composition.

9. R. R. Chauvette, P. A. Pennington, C. W. Ryan, R. D. G. Cooper, F. L. José, I. G. Wright, E. M. Van Heyningen and G. W. Huffman, J. Org. Chem., **36**, 1259 (1971).
10. B. M. Trost and T. R. Verhoeven, J. Org. Chem., **41**, 3215 (1976). Intermolecular allylic alkylations using disubstituted olefins results in partial loss of olefin stereochemistry; T. R. Verhoeven, unpublished results.
11. B. M. Trost, H. C. Arndt, P. E. Strege and T. R. Verhoeven, Tetrahedron Letters, 3477 (1976).
12. We are grateful to Prof. H. Gerlach, Laboratorium für Organische Chemie der Eidgenössischen Technischen Hochschule, Zurich, for the reference sample and to Prof. E. J. Corey, Harvard University, Cambridge, Mass., for spectral data.